

## STRUCTURAL PHASIC CHARACTERISTICS OF NANOPARTICLES OXIDES SAMARIUM-153 BY INFLUENCE GAMMA ACTIVATION

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The isotope  $^{153}\text{Sm}$  ( $T_{1/2} = 46.3$  h,  $E = 810$  keV) which emit is both therapeutic beta and diagnostic gamma energies and it is ideal for therapy plus diagnostic application. The specific activity of  $^{153}\text{Sm}$  was increased by using the Szilard-Chalmers reaction and nanoparticles  $\text{Sm}_2\text{O}_3$  after gamma activation on LAE. It is shown that the isotope  $^{153}\text{Sm}$  after gamma activation there are no radioactive impurities and no changes of chemical and phasic structure.

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### INTRODUCTION

Samarium-153 ( $^{153}\text{Sm}$ ) has radiation characteristics such as the medium-energy beta particle emission ( $E_{\text{max}} = 810$  keV) which is profitable for treatment, the medium-energy gamma quantum (103 keV) which is suitable for imaging, and the short half-life (46.3 h). Isotopes which emit therapeutic beta and diagnostic gamma energies would be ideal for application – “theranostic”. An ideal theranostic radionuclide should have optimum physical half-life, necessary linear energy and range in tissues (1 cm), high ratio of non-penetrating to penetrating, short lived or stable daughter, good and selective concentration with prolonged retention in tumor, and minimum uptake by normal tissue [1].

Samarium-153 is potential suitable as an alternative to  $^{90}\text{Y}$ ,  $^{32}\text{P}$ ,  $^{90}\text{Sr}$  in different cancer treatment. It has been widely used for palliative pain treatment bone metastatic patients but its therapeutic potential has not been fully utilized for other cases. The imaging properties of  $^{153}\text{Sm}$  have been proved effective for gastrointestinal scintigraphy [2].

The main advantages of  $^{153}\text{Sm}$  are the important features for transarterial radioembolization.  $^{153}\text{Sm}$  is the major therapeutic agent which is widely used in the world just for various bone pain palliative therapy [3].

One of the important features of radioembolic agent is the particle size. Nowadays the particular attention is given to the particles size especially nanoparticles in the range between 40...80 nm. The nanoparticles may pass in the tumor capillaries and reach the such parenchymatous organs such as lungs, kidneys, liver [4].

Nanoparticles with resistivity to physical heat and body chemicals, biocompatible, non-biodegradable and easily labeled with radionuclides are highly preferred.

Decay properties such as half-life and particles energy play significant roles in clinical characteristics, for example, duration of palliative effects and degree of and time to recovery from myelosuppression.

The particles emissions from  $^{32}\text{P}$  and  $^{89}\text{Sr}$  in bone and soft tissue are much greater than those of  $^{153}\text{Sm}$ . High energy particles are associated with greater marrow toxicity as a result of the larger volumes of marrow exposed to radiation. The shorter physical half-life of  $^{153}\text{Sm}$  (1.9 days) results in more rapid delivery of

radiation than either  $^{32}\text{P}$  (14.3 days) or  $^{89}\text{Sr}$  (50.5 days). For example, delivery of 90% of the total dose of radiation requires approximately 3.5 half-lives of decay, a time interval of approximately 1 week for  $^{153}\text{Sm}$ .

The purpose of our study was to evaluate for each sample  $^{153}\text{Sm}$  the presence of radionuclide impurities, especially the long-lived radionuclides and comparison of IR spectra nanoparticles initial  $\text{Sm}_2\text{O}_3$  and  $^{153}\text{Sm}$  on the water content before and after gamma activation.

### 1. MATERIALS AND METHODS

Gamma-activation method used for production  $^{153}\text{Sm}$  with high specific activity that can be increased in many cases by using the Szilard-Chalmers process [5–7].

In that experiment the nanoparticles (50...80 nm)  $\text{Sm}_2\text{O}_3$  and clinoptilolite (80 nm) were used. On LAE with  $E = 23$  MeV and current 700  $\mu\text{A}$  it is possible to produce 1 Ci  $^{153}\text{Sm}$  during day by using of  $\text{Sm}_2\text{O}_3$  (40 g) of natural isotope composition.

The activity of  $^{153}\text{Sm}$  obtained in nuclear reaction  $^{154}\text{Sm}(\gamma, n)^{153}\text{Sm}$  measured by Ge(Li)-detector with volume 50  $\text{cm}^3$  and with energy resolution 3.2 keV in the area of 1332. To reduce the influence of background, the detector is equipped with a three-layer Pb-Cu-Al protection.

Nanoparticles  $\text{Sm}_2\text{O}_3$  and clinoptilolite used as donor and acceptor, respectively. For the concentration of recoil nuclei in among donor, nanoparticles sizes  $\text{Sm}_2\text{O}_3$ , containing an activatable element, must be less than or equal to the range of recoil nuclei.

The structure and phase composition of nanoparticles  $\text{Sm}_2\text{O}_3$  and  $^{153}\text{Sm}$  investigated by infrared (IR) spectrophotometer “Specord-75” in the frequency 600...4000  $\text{cm}^{-1}$  on the compressed tablets from mixture KBr (100 mg sample) [8].

### 2. RESULTS AND DISCUSSION

Gamma ray spectrum of  $^{153}\text{Sm}$  is shown in Fig. 1.

The most dominant peaks observed were at 103 and 69.5 keV and two other significant peaks were 41 and 45 keV. These peaks resulted from K-shell characteristic X-ray following radioactive decay. There are not radionuclide impurities in the  $^{153}\text{Sm}$  sample.

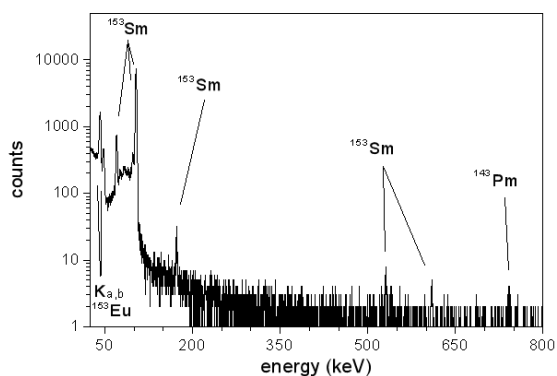


Fig. 1. Gamma ray spectrum of natural Sm after activation bremsstrahlung energy 11.5 MeV

Gamma ray spectrum of native non activate clinoptilolite as acceptor for  $^{153}\text{Sm}$  is shown in Fig. 2.

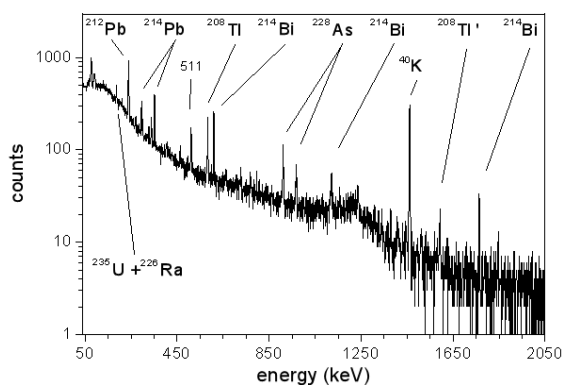


Fig. 2. Gamma ray spectrum of natural clinoptilolite

The measurement of gamma spectra of natural clinoptilolite gamma spectrum was shown the following gamma lines of  $^{228}\text{Ac}$  and  $^{40}\text{K}$  for  $^{214}\text{Pb}$ ,  $^{214}\text{Bi}$ ,  $^{212}\text{Pb}$ ,  $^{208}\text{Tl}$ . The content of Th, U was conformed with distribution theirs in earth's crust (Th  $\sim 1.0 \cdot 10^{-5}$ , U  $\sim 3.6 \cdot 10^{-6}$  g/g). The ratio Th/U was 3.87 for natural clinoptilolite sample and for earth's crust of average value – 2.78.

IR spectrum of the nanoparticle sample  $\text{Sm}_2\text{O}_3$  was compared to IR spectrum of the same sample after activation by bremsstrahlung energy 11.5 MeV. As shown in Figs. 3 and 4. IR spectra of samples before and after activation are nearly identical. Gamma activation did not destroy the chemical structure and the functional groups. The intensity bands 3200 and 3400  $\text{cm}^{-1}$  also indicated that the water content during gamma activation of samples was not changed.

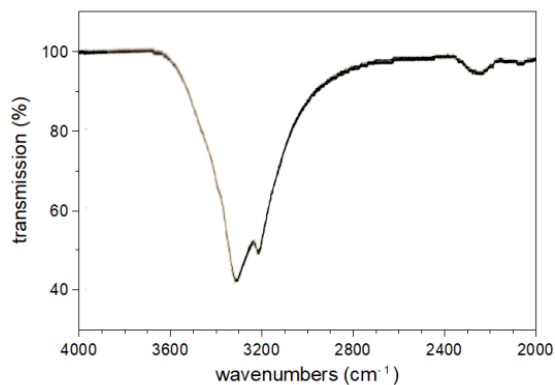


Fig. 3. IR spectra of nanoparticles  $\text{Sm}_2\text{O}_3$  before activation bremsstrahlung energy 11.5 MeV

The presence in the IR spectrum of absorption bands of medium intensity in the frequency range of 2000 and 2800  $\text{cm}^{-1}$  small peaks which is associated with an impurity  $\text{OH}^{-1}$  groups.

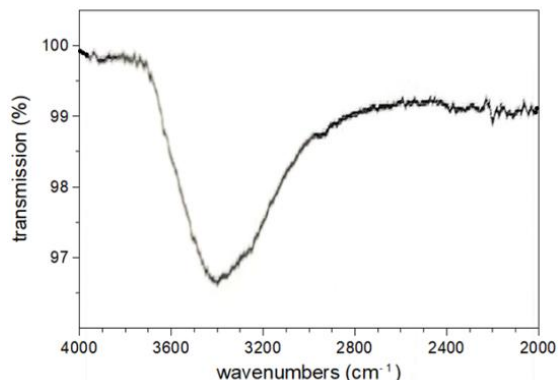


Fig. 4. IR spectra of nanoparticles  $\text{Sm}_2\text{O}_3$  after activation bremsstrahlung energy 11.5 MeV

## CONCLUSIONS

Photo-nuclear technology for produce isotopes  $^{153}\text{Sm}$  with high specific activity by using the Szilard-Chalmers process and nanoparticles  $\text{Sm}_2\text{O}_3$  was elaborated. Elemental analysis and method of infrared spectroscopy showed no radioactive impurities and no major differences observed between chemical structure, functional groups and water content after gamma activation of nanoparticles  $\text{Sm}_2\text{O}_3$ .

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## ВПЛИВ ГАММА-АКТИВАЦІЇ НА СТРУКТУРНО-ФАЗОВІ ХАРАКТЕРИСТИКИ НАНОЧАСТИНОК $^{153}\text{Sm}$

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Ізотоп  $^{153}\text{Sm}$  ( $T_{1/2} = 46.3$  год,  $E = 810$  кеВ) являється ідеальним емітером для проведення бета-терапії та гамма-діагностики. Питома активність  $^{153}\text{Sm}$  була підвищена завдяки використанню реакції Сціларда-Чалмерса та наночастинок  $\text{Sm}_2\text{O}_3$  після гамма-активації на лінійному прискорювачі електронів. Показано, що ізотоп  $^{153}\text{Sm}$  після гамма-активації не містить радіонуклідних домішок та не втрачає хімічну та фазову структури.